# Generation and Diels-Alder Reactions of 3-Acyl-3-pyrroline-2,5-diones 

Michael L. Durrant and Eric J. Thomas*<br>The Dyson Perrins Laboratory, South Parks Road, Oxford OX1 30Y


#### Abstract

A series of 3 -acyl-3-pyrroline-2,5-diones (13)-(17) have been generated in solution by oxidative elimination from the corresponding 3 -acyl-3-phenylselenopyrrolidine-2,5-diones (8)-(12), and trapped by the conjugated dienes, trans,trans-hexa-2,4-diene and cyclopentadiene, to give modest to good yields of the Diels-Alder adducts (18)-(27). The imido ring dominated the endo-exo selectivity.


The use of the Diels-Alder reaction as a synthetic method is still being developed. Recently some Diels-Alder reactions of the methoxycarbonylmaleic anhydride (1) were described. ${ }^{1}$ We here report the generation in solution of a series of related dienophiles, namely 3 -acyl-3-pyrroline-2,5-diones (13)-(17), which were studied because of their possible use in a proposed cytochalasan synthesis, ${ }^{2}$ and which were found to undergo rapid Diels-Alder reactions with simple conjugated dienes.

## Results and Discussion

1-Methylpyrrolidine-2,5-dione (2) was acylated at the 3position, following the published procedure, using an ester and sodium hydride in tetrahydrofuran, in the presence of a catalytic amount of an alcohol, either methanol or ethanol. ${ }^{3}$ The 3 -acylpyrrolidine-2,5-diones (3)-(7) so prepared were found to exist predominantly in their keto-forms; enolisation occurred to a small extent (ca. $10 \%$ ), as shown by ${ }^{1} \mathrm{H}$ n.m.r. spectroscopy. Benzeneselenenylation was then achieved by treatment of the 3 -acylpyrrolidine-2,5-diones with sodium hydride and benzeneselenenyl chloride to give the corresponding 3 -phenylseleno products (8)-(12). ${ }^{4}$ These selenides were oxidised using two equivalents of $m$-chloroperoxybenzoic acid in dichloromethane at $-15^{\circ} \mathrm{C}$. After 15 min , excess of pyridine was added, followed by an excess of either cyclopentadiene or trans,trans-hexa-2,4-diene. The Diels-Alder adducts (18)-(27) were then isolated after the mixture had been stirred for $1-2 \mathrm{~h}$ at room temperature. The products obtained, together with their yields, are given in the Scheme.

The Diels-Alder adducts were identified on the basis of their spectroscopic data. For each of the cyclopentadiene reactions, the high field ( 300 MHz ) ${ }^{1} \mathrm{H}$ n.m.r. spectrum of the crude product mixture showed the presence of two products in the ratio $c a .85: 15$. In all cases the major product was isolated pure, and identified as the adduct with the stereochemistry shown in

Figure 1, i.e. with the imido ring in the endo position. This stereochemistry was established by spin-decoupling experiments which showed the presence of coupling, $J c a .4 .6 \mathrm{~Hz}$, between the exo proton, $6-\mathrm{H}$, and the bridgehead proton, $7-\mathrm{H}$. This coupling, together with the absence of long-range coupling between $6-\mathrm{H}$ and the anti-methylene bridge proton, $10^{\prime}-\mathrm{H}$, established that the stereochemistry is as shown. ${ }^{5}$ The minor product from each of the cyclopentadiene reactions was assumed to be the Diels-Alder adduct with the imido ring exo. The ${ }^{1} \mathrm{H}$ n.m.r. data obtained were consistent with this, but it was not possible to isolate pure samples of these minor products for proper characterisation.

For the trans,trans-hexa-2,4-diene adducts (23)-(27), only a single Diels-Alder adduct was detected in each case, identified as the adduct with the imido ring endo as shown in Figure 2. This stereochemical assignment was made by analogy with the cyclopentadiene adducts discussed above, and is consistent with the $\mathrm{H}(5)-\mathrm{H}(6)$ coupling constants, $J 5-5.7 \mathrm{~Hz}$, which are in the range expected for such systems. ${ }^{2,6}$ These adducts would appear to adopt the boat conformation shown. Thus, the vinylic protons give rise to a symmetric AB system of triplets in their ${ }^{1}$ H n.m.r. spectra.
These observations show that 3-acylpyrroline-2,5-diones can be generated and trapped as Diels-Alder adducts. Preliminary attempts were made to isolate these intermediate dienophiles but were unsuccessful, only polymeric material being obtained.

(1)


Scheme. * The pyrrolinediones (13)-(17) were not isolated.


Figure 1.


Figure 2.

The yields of the Diels-Alder adducts from the 3-aroylpyrroline-2,5-diones (13)-(15) were quite reasonable, being in the range $50-75 \%$. However, lower yields were obtained from the aliphatic analogues (16) and (17), possibly because of competing polymerisation initiated by enolisation. The speed of the Diels-Alder reactions of the acylpyrroline-2,5-diones, all of which were complete in less than 2 h at room temperature, is consistent with the electron deficient nature of the double bond in these systems. ${ }^{7}$ Since there exists considerable synthetic methodology based on reactions of imides, these Diels-Alder reactions may prove to be useful in synthesis.

## Experimental

I.r. spectra were measured on Perkin-Elmer 257 and 297 spectrophotometers, and ${ }^{1} \mathrm{H}$ n.m.r. spectra on a Bruker WH300 spectrometer ( 300 MHz ). M.p.s were determined on a Buchi 510 apparatus, and are uncorrected. Mass spectra were measured on a VG-micromass ZAB-16F spectrometer using either electron impact (E.I.) or chemical ionization (C.I.) modes.
T.l.c. was carried out using Merck aluminium sheets, precoated with silica gel $60 \mathrm{~F}^{\prime}{ }_{254}$; flash chromatography was on Merck silica gel 60 , and short column chromatography used Merck Keiselgel 60H.

All solvents were dried and distilled before use. Ether refers to diethyl ether throughout and light petroleum to the fraction with b.p. $40-60^{\circ} \mathrm{C}$. 1-Methylpyrrolidine-2,5-dione (2) was prepared by treatment of succinic anhydride with methylamine in aqueous ethanol, ${ }^{8}$ m.p. $64{ }^{\circ} \mathrm{C}$ (from isopropyl alcohol) (lit., ${ }^{9}$ $66-68^{\circ} \mathrm{C}$ ).

General Procedure for the Acylation of 1-Methylpyrrolidine-2,5-dione (2).-The ester ( 2 mol equiv.) and a few drops of methanol (for methyl esters) or ethanol (for ethyl esters), were added to 1 -methylpyrrolidine-2,5-dione (2) in anhydrous tetrahydrofuran (THF). Sodium hydride ( $60 \%$ dispersion in oil, 2 mol equiv.) was added in small portions, and the heterogeneous mixture heated under reflux for $16-20 \mathrm{~h}$, cooled, and ether and water were added. The aqueous phase was separated, acidified with glacial acetic acid, and extracted into dichloromethane. After being dried $\left(\mathrm{MgSO}_{4}\right)$, the organic extract was concentrated under reduced pressure, and the product isolated by crystallisation or distillation. The following compounds were prepared. 1-Methyl-3-(4-methylbenzoyl)pyrrol-idine-2,5-dione (3) ( $46 \%$ ), m.p. $107-109{ }^{\circ} \mathrm{C}$ (from methanol) (Found: C, 67.4; H, 5.55; N, 6.05. $\mathrm{C}_{13} \mathrm{H}_{13} \mathrm{NO}_{3}$ requires C, 67.5; H, 5.65 ; N, $6.05 \%$ ); $v_{\text {max }}\left(\mathrm{CH}_{2} \mathrm{Cl}_{2}\right) 3060$ (aromatic CH ), 1780 , 1710 , and $1680(\mathrm{C}=\mathrm{O})$, and $1610 \mathrm{~cm}^{-1}$ (aromatic ring); $\delta_{\mathrm{H}}\left(\mathrm{CDCl}_{3}\right) 2.45(3 \mathrm{H}, \mathrm{s}$, aromatic Me$), 2.85(1 \mathrm{H}, \mathrm{dd}, J 18,9 \mathrm{~Hz}$, 4-H), 3.01 ( $3 \mathrm{H}, \mathrm{s}, \mathrm{NMe}$ ), 3.39 ( $1 \mathrm{H}, \mathrm{dd}, J 18,4 \mathrm{~Hz}, 4-\mathrm{H}$ ), 4.83 ( 1 $\mathrm{H}, \mathrm{dd}, J 9,4 \mathrm{~Hz}, 3-\mathrm{H}$ ), and 7.33 and 8.00 (each $2 \mathrm{H}, \mathrm{m}$, aromatic H); $m / z$ (E.I.) $231\left(M^{+}, 6 \%\right)$ and $119\left(M^{+}-112,100\right)$.

3-Benzoyl-1-methylpyrrolidine-2,5-dione (4) (64\%), m.p. $100-$ $103^{\circ} \mathrm{C}$ (from methanol) (Found: C, $66.15 ; \mathrm{H}, 5.15 ; \mathrm{N}, 6.6$. $\mathrm{C}_{12} \mathrm{H}_{11} \mathrm{NO}_{3}$ requires $\mathrm{C}, 66.35 ; \mathrm{H}, 5.1 ; \mathrm{N}, 6.45 \%$ ); $v_{\text {max. }}\left(\mathrm{CH}_{2} \mathrm{Cl}_{2}\right.$ ) 3060 (aromatic CH ), 1785 and $1700(\mathrm{C}=\mathrm{O})$, and $1600 \mathrm{~cm}^{-1}$ (aromatic ring); $\delta_{\mathrm{H}}\left(\mathrm{CDCl}_{3}\right) 2.86(1 \mathrm{H}$, dd, $J 18,9 \mathrm{~Hz}, 4-\mathrm{H})$, $3.00(3 \mathrm{H}, \mathrm{s}, \mathrm{NMe}), 3.37$ ( $1 \mathrm{H}, \mathrm{dd}, J 18,4 \mathrm{~Hz}, 4-\mathrm{H}$ ), 4.85 ( $1 \mathrm{H}, \mathrm{dd}, J$
$9,4 \mathrm{~Hz}, 3-\mathrm{H}), 7.58(3 \mathrm{H}, \mathrm{m}$, aromatic H$)$, and $8.12(2 \mathrm{H}, \mathrm{m}$, aromatic H); $m / z$ (E.I.) $217\left(M^{+}, 7 \%\right)$ and $105\left(M^{+}-112,100\right)$.

3-(4-Methoxybenzoyl)-1-methylpyrrolidine-2,5-dione (5) ( $25 \%$ ), m.p. 119- $120.5^{\circ} \mathrm{C}$ (from methanol) (Found: C, 63.1; H, 5.3; $\mathrm{N}, 5.7 . \mathrm{C}_{13} \mathrm{H}_{13} \mathrm{NO}_{4}$ requires $\mathrm{C}, 63.15 ; \mathrm{H}, 5.3 ; \mathrm{N}, 5.65 \%$ ); $v_{\text {max. }}\left(\mathrm{CH}_{2} \mathrm{Cl}_{2}\right) 1780,1705$, and $1675(\mathrm{C}=\mathrm{O})$, and $1600 \mathrm{~cm}^{-1}$ (aromatic); $\delta_{\mathrm{H}}\left(\mathrm{CDCl}_{3}\right) 2.82(1 \mathrm{H}, \mathrm{dd}, J 18,9 \mathrm{~Hz}, 4-\mathrm{H}), 2.97(3 \mathrm{H}$, s, NMe), 3.37 ( $1 \mathrm{H}, \mathrm{dd}, J 18,4 \mathrm{~Hz}, 4-\mathrm{H}$ ), 3.89 ( $3 \mathrm{H}, \mathrm{s}$, OMe), 4.78 $(1 \mathrm{H}, \mathrm{dd}, J 9,4 \mathrm{~Hz}, 3-\mathrm{H}$ ), and 6.99 and 8.08 (each $2 \mathrm{H}, \mathrm{m}$, aromatic H); $m / z$ (E.I.) $247\left(M^{+}, 4 \%\right), 152\left(M^{+}-95,61\right)$, and 135 ( $M^{+}-112,100$ ).

3-(1-Oxopropyl)-1-methylpyrrolidine-2,5-dione (6) (58\%), b.p. $200-210^{\circ} \mathrm{C}$ at 0.9 mmHg (Found: $\mathrm{C}, 56.65 ; \mathrm{H}, 6.25 ; \mathrm{N}, 8.2$. $\mathrm{C}_{8} \mathrm{H}_{11} \mathrm{NO}_{3}$ requires $\mathrm{C}, 56.8 ; \mathrm{H}, 6.55 ; \mathrm{N}, 8.3 \%$ ); $v_{\text {max. }}$ (liq. film) 1780 and $1710 \mathrm{~cm}^{-1}(\mathrm{C}=0) ; \delta_{\mathrm{H}}\left(\mathrm{CDCl}_{3}\right) 1.11(3 \mathrm{H}, \mathrm{t}, J 7.5 \mathrm{~Hz}$, $\left.\mathrm{CH}_{2} \mathrm{CH}_{3}\right), 2.64\left(1 \mathrm{H}, \mathrm{dq}, J 19,7.5 \mathrm{~Hz}, \mathrm{HCHCH}_{3}\right), 2.67(1 \mathrm{H}, \mathrm{dd}$, $J 20,10 \mathrm{~Hz}, 4-\mathrm{H}), 2.97(3 \mathrm{H}, \mathrm{s}, \mathrm{NMe}), 3.14(1 \mathrm{H}, \mathrm{dq}, J 19,7.5 \mathrm{~Hz}$, $\left.\mathrm{HCHCH}_{3}\right), 3.32(1 \mathrm{H}, \mathrm{dd}, J 20,5 \mathrm{~Hz}, 4-\mathrm{H})$, and $3.95(1 \mathrm{H}, \mathrm{dd}, J$ $10,5 \mathrm{~Hz}, 3-\mathrm{H}) ; m / z$ (E.I.) $169\left(M^{+}, 9 \%\right) 113\left(M^{+}-56,41\right)$, and $57\left(M^{+}-112,100\right)$.

1-Methyl-3-(2-methyl-1-oxopropyl)pyrrolidine-2,5-dione (7) $\left(24 \%\right.$ ), b.p. $175-180^{\circ} \mathrm{C}$ at 0.7 mmHg (Found: C, 58.95 ; H, 7.3; $\mathrm{N}, 7.65 . \mathrm{C}_{9} \mathrm{H}_{13} \mathrm{NO}_{3}$ requires $\mathrm{C}, 59.0 ; \mathrm{H}, 7.15 ; \mathrm{N}, 7.65 \%$ ); $v_{\text {max. }}$ (liq. film) 1780 and $1710 \mathrm{~cm}^{-1}(\mathrm{C}=\mathrm{O}) ; \delta_{\mathrm{H}}\left(\mathrm{CDCl}_{3}\right) 1.13$ and 1.20 (each $3 \mathrm{H}, \mathrm{d}, J 7 \mathrm{~Hz}, \mathrm{CH} M e_{2}$ ), $2.67(1 \mathrm{H}, \mathrm{dd}, J 18,9 \mathrm{~Hz}, 4-$ H), 2.96 ( $3 \mathrm{H}, \mathrm{s}, \mathrm{NMe}$ ), 3.21 ( $2 \mathrm{H}, \mathrm{m}, \mathrm{CHMe} \mathrm{C}_{2}+4-\mathrm{H}$ ), and 4.13 ( $1 \mathrm{H}, \mathrm{dd}, J 9,4 \mathrm{~Hz}, 3-\mathrm{H}$ ); $m / z$ (E.I.) $183\left(M^{+}, 8 \%\right)$ and $71\left(M^{+}-\right.$ 112,40 ).

General Procedure for the Benzeneselenenylation of 3-Acyl-1-methylpyrrolidine-2,5-diones.-The 3-acyl-1-methylpyrrolidine2,5 -dione ( 1 mol equiv.) was added to a suspension of sodium hydride ( 1.2 mol equiv., $60 \%$ dispersion in oil, washed $3 \times$ with light petroleum) in anhydrous THF at $0^{\circ} \mathrm{C}$, and the mixture allowed to warm to $c a .20^{\circ} \mathrm{C}$ and stirred for 30 min . After recooling to $0^{\circ} \mathrm{C}$, benzeneselenenyl chloride ( 1.1 mol equiv.) in anhydrous THF was added, and the mixture again allowed to warm to $\mathrm{ca} .20^{\circ} \mathrm{C}$ and stirred for 30 min . After the addition of ether and saturated aq. $\mathrm{NaHCO}_{3}$ the ethereal phase was separated, dried $\left(\mathrm{MgSO}_{4}\right)$, and concentrated under reduced pressure to give the impure products which were crystallised or purified by flash chromatography (eluted with $1: 1$ light petroleum-ether). The following compounds were so prepared. 1-Methyl-3-(4-methylbenzoyl)-3-phenylselenopyrrolidine-2,5-
dione (8) $(33 \%)$, m.p. $152-154{ }^{\circ} \mathrm{C}$ (from benzene-methanol); $v_{\text {max. }}\left(\mathrm{CH}_{2} \mathrm{Cl}_{2}\right) 1780,1710$, and $1670(\mathrm{C}=0)$, and $1610 \mathrm{~cm}^{-1}$ (aromatic); $\delta_{\mathrm{H}}\left(\mathrm{CDCl}_{3}\right) 2.44(3 \mathrm{H}, \mathrm{s}$, aromatic Me$), 2.78(3 \mathrm{H}, \mathrm{s}$, NMe), 3.11 and 3.77 (each $1 \mathrm{H}, \mathrm{d}, J 19 \mathrm{~Hz}, \mathrm{CH}_{2}$ ), $7.4(7 \mathrm{H}, \mathrm{m}$, aromatic H), and $7.99(2 \mathrm{H}, \mathrm{m}$, aromatic H); $m / z$ (E.I.) 385,387 $\left(M^{+}, 0.7 \%\right)$ and $119\left(M^{+}-266,268,100\right)$.

3-Benzoyl-1-methyl-3-phenylselenopyrrolidine-2,5-dione (9) ( $59 \%$ ), m.p. $152-154{ }^{\circ} \mathrm{C}$ (from benzene) (Found: C, $58.3 ; \mathrm{H}$, $4.05 ; \mathrm{N}, 3.95 . \mathrm{C}_{18} \mathrm{H}_{15} \mathrm{NO}_{3} \mathrm{Se}$ requires C, $58.1 ; \mathrm{H}, 4.05 ; \mathrm{N}, 3.75 \%$ ); $v_{\text {max. }}\left(\mathrm{CH}_{2} \mathrm{Cl}_{2}\right) 3060($ aromatic H$), 1780,1710,1680(\mathrm{C}=\mathrm{O})$, and $1595 \mathrm{~cm}^{-1}$ (aromatic ring); $\delta_{\mathrm{H}}\left(\mathrm{CDCl}_{3}\right) 2.79(3 \mathrm{H}, \mathrm{s}, \mathrm{NMe})$, 3.11 and 3.78 (each $\left.1 \mathrm{H}, \mathrm{d}, J 19 \mathrm{~Hz}, \mathrm{CH}_{2}\right), 7.5(8 \mathrm{H}, \mathrm{m}$, aromatic H ), and $8.10\left(2 \mathrm{H}, \mathrm{m}\right.$, aromatic H); $m / z$ (C.I.) $372,374\left(M^{+}+1\right.$, $4 \%$ ).

3-(4-Methoxybenzoyl)-1-methyl-3-phenylselenopyrrolidine2,5 -dione (10) $\left(28 \%\right.$ ), m.p. $119-121^{\circ} \mathrm{C}$ (from benzene-ether); $v_{\text {max. }}\left(\mathrm{CH}_{2} \mathrm{Cl}_{2}\right) 1780,1710$, and $1665(\mathrm{C}=\mathrm{O})$, and 1600 and $1580 \mathrm{~cm}^{-1}$ (aromatic ring); $\delta_{\mathrm{H}}\left(\mathrm{CDCl}_{3}\right) 2.78(3 \mathrm{H}, \mathrm{s}, \mathrm{NMe})$, 3.10 and 3.83 (each $\left.1 \mathrm{H}, \mathrm{d}, J 19 \mathrm{~Hz}, \mathrm{CH}_{2}\right), 3.89(3 \mathrm{H}$, s, OMe), $6.97(2 \mathrm{H}, \mathrm{m}$, aromatic H$), 7.45(5 \mathrm{H}, \mathrm{m}$, aromatic H$)$, and 8.17 ( $2 \mathrm{H}, \mathrm{m}$, aromatic H); $m / z$ (E.I.) 401, $403\left(M^{+}, 3 \%\right.$ ), 245 $\left(M^{+}-156,158,53\right), 155$ and $157\left(M^{+}-246,48\right)$, and $135\left(M^{+}\right.$ $-266,268,100)$.

3-(1-Oxopropyl)-1-methyl-3-phenylselenopyrrolidine-2,5dione (11) $\left(74 \%\right.$ ), m.p. $62-64{ }^{\circ} \mathrm{C}$ (from ether-benzene) (Found: $\mathrm{C}, 51.55 ; \mathrm{H}, 4.85 ; \mathrm{N}, 4.6 . \mathrm{C}_{14} \mathrm{H}_{15} \mathrm{NO}_{3} \mathrm{Se}$ requires $\mathrm{C}, 51.85 ; \mathrm{H}$, $4.65 ; \mathrm{N}, 4.3 \%) ; v_{\text {max. }}\left(\mathrm{CH}_{2} \mathrm{Cl}_{2}\right) 1780$ and $1710 \mathrm{~cm}^{-1}(\mathrm{C}=\mathrm{O})$; $\delta_{\mathrm{H}}\left(\mathrm{CDCl}_{3}\right) 1.15\left(3 \mathrm{H}, \mathrm{t}, J 7.3 \mathrm{~Hz}, \mathrm{CH}_{2} \mathrm{CH}_{3}\right), 2.69(1 \mathrm{H}, \mathrm{d}, J 19.5$ $\mathrm{Hz}, 4-\mathrm{H}), 2.88(3 \mathrm{H}, \mathrm{s}, \mathrm{NMe}), 2.96$ and 3.34 (each $1 \mathrm{H}, \mathrm{dq}, J 18.5$, $7 \mathrm{~Hz}, \mathrm{CH}_{2} \mathrm{CH}_{3}$ ), $3.56(1 \mathrm{H}, \mathrm{d}, J 19.5 \mathrm{~Hz}, 4-\mathrm{H})$, and $7.4(5 \mathrm{H}, \mathrm{m}$, aromatic H); $m / z$ (E.I.) $325,323\left(M^{+}, 6 \%\right)$.

1-Methyl-3-(2-methyl-1-oxopropyl)-3-phenylselenopyrrol-idine-2,5-dione (12) $\left(56 \%\right.$ ), m.p. $64-66^{\circ} \mathrm{C}$ (from benzene-light petroleum) (Found: C, $53.5 ; \mathrm{H}, 5.1 ; \mathrm{N}, 4.25 . \mathrm{C}_{15} \mathrm{H}_{17} \mathrm{NO}_{3} \mathrm{Se}$ requires $\mathrm{C}, 53.25 ; \mathrm{H}, 5.05 ; \mathrm{N}, 4.15 \%)$; $v_{\text {max. }}\left(\mathrm{CH}_{2} \mathrm{Cl}_{2}\right) 1780$ and $1710 \mathrm{~cm}^{-1}(\mathrm{C}=0) ; \delta_{\mathrm{H}}\left(\mathrm{CDCl}_{3}\right) 1.02$ and $1.36($ each $3 \mathrm{H}, \mathrm{d}, J 5 \mathrm{~Hz}$, $\mathrm{HCMe}_{2}$ ), $2.66(1 \mathrm{H}, \mathrm{d}, J 20 \mathrm{~Hz}, 4-\mathrm{H}), 2.93$ ( $3 \mathrm{H}, \mathrm{s}$, NMe), 3.49 ( 1 $\mathrm{H}, \mathrm{d}, J 20 \mathrm{~Hz}, 4-\mathrm{H}), 3.87\left(1 \mathrm{H}, \mathrm{h}, J 5 \mathrm{~Hz}, H \mathrm{CMe}_{2}\right)$, and $7.43(5 \mathrm{H}$, m , aromatic H); $m / z 337,339\left(M^{+}, 4 \%\right)$.

Generation and Diels-Alder Reactions of 3-Acyl-3-pyrroline-2,5-diones (13)-(17).- $m$-Chloroperoxybenzoic acid ( 2 mol equiv.) was added to the phenylselenopyrrolidine-2,5-dione in anhydrous dichloromethane at $-15^{\circ} \mathrm{C}$. After the mixture had been stirred for $10-15 \mathrm{~min}$ at $-15^{\circ} \mathrm{C}$, pyridine $(3.1 \mathrm{~mol}$ equiv.) and the diene ( 10 mol equiv.) were added, and the mixture was allowed to warm to ca. $20^{\circ} \mathrm{C}$ and stirred for 1 h (using cyclopentadiene) and for 2 h (using trans, trans-hexa-2,4diene). After the addition of 1 m -aqueous NaOH and more $\mathrm{CH}_{2} \mathrm{Cl}_{2}$, the organic phase was separated, dried ( $\mathrm{MgSO}_{4}$ ), and concentrated under reduced pressure. The products were purified by short column chromatography (eluted with light petroleum-ether). The ${ }^{1} \mathrm{H}$ n.m.r. spectra of the crude products from the cyclopentadiene reactions indicated the presence of two components in each case, ratio ca. 85:15. The major component was isolated pure (see text). The following adducts were characterised. (1RS,2SR,6RS,7SR)-4-Methyl-2-(4-methyl-benzoyl)-3,5-dioxo-4-azatricyclo[5.2.1.0 ${ }^{2.6}$ ]dec-8-ene (18) (67\%), m.p. $127-128^{\circ} \mathrm{C}$ (from ether-benzene) (Found: C, 72.85 ; H, 5.9; $\mathrm{N}, 4.65 . \mathrm{C}_{18} \mathrm{H}_{17} \mathrm{NO}_{3}$ requires $\mathrm{C}, 73.2 ; \mathrm{H}, 5.8 ; \mathrm{N}, 4.75 \%$ ); $v_{\text {max }}\left(\mathrm{CH}_{2} \mathrm{Cl}_{2}\right) 3020$ (aromatic H ), 1770, 1705 , and 1670 $(\mathrm{C}=\mathrm{O})$, and $1610 \mathrm{~cm}^{-1}$ (aromatic ring); $\delta_{\mathrm{H}}\left(\mathrm{CDCl}_{3}\right) 1.53(1 \mathrm{H}, \mathrm{d}$, $J 9 \mathrm{~Hz}, 10-\mathrm{H}), 1.72(1 \mathrm{H}, \mathrm{dt}, J 9,1.6 \mathrm{~Hz}, 10-\mathrm{H}), 2.42(3 \mathrm{H}, \mathrm{s}$, aromatic Me), $2.88(3 \mathrm{H}, \mathrm{s}, \mathrm{NMe}), 3.41$ and 3.88 (each $1 \mathrm{H}, \mathrm{m}, 7-$ H and $1-\mathrm{H}), 4.01(1 \mathrm{H}, \mathrm{d}, J 4.7 \mathrm{~Hz}, 6-\mathrm{H}), 6.27(2 \mathrm{H}, \mathrm{m}$, vinylic H$)$, and 7.32 and 8.06 (each $2 \mathrm{H}, \mathrm{m}$, aromatic H); $m / z$ (E.I.) 295 ( $M^{+}$, $2 \%$ ) and $119\left(M^{+}-176,100\right)$.
(1RS,2SR,6RS,7SR)-2-Benzoyl-4-methyl-3,5-dioxo-4-azatricyclo[5.2.1.0 ${ }^{2.6}$ ]dec-8-ene (19) ( $74 \%$ ), m.p. $95-98{ }^{\circ} \mathrm{C}$ (from ether) (Found: $\mathrm{C}, 72.45 ; \mathrm{H}, 5.25 ; \mathrm{N}, 5.05 . \mathrm{C}_{17} \mathrm{H}_{15} \mathrm{NO}_{3}$ requires $\mathrm{C}, 72.6 ; \mathrm{H}, 5.35 ; \mathrm{N}, 5.0 \%$ ); $v_{\text {max. }}\left(\mathrm{CH}_{2} \mathrm{Cl}_{2}\right) 3060$ (aromatic H), 1775 and $1705(\mathrm{C}=\mathrm{O})$, and 1600 and $1580 \mathrm{~cm}^{-1}$ (aromatic ring); $\delta_{\mathrm{H}}\left(\mathrm{CDCl}_{3}\right) 1.53(1 \mathrm{H}, \mathrm{d}, J 9 \mathrm{~Hz}, 10-\mathrm{H}), 1.74(1 \mathrm{H}, \mathrm{dt}, J 9,1.6$ $\mathrm{Hz}, 10-\mathrm{H}$ ), 2.89 ( $3 \mathrm{H}, \mathrm{s}, \mathrm{NMe}$ ), 3.42 and 3.88 (each $1 \mathrm{H}, \mathrm{m}, 7-\mathrm{H}$ and $1-\mathrm{H}), 4.05(1 \mathrm{H}, \mathrm{d}, J 4.6 \mathrm{~Hz}, 6-\mathrm{H}), 6.29(2 \mathrm{H}, \mathrm{m}$, vinylic H$)$, $7.55(3 \mathrm{H}, \mathrm{m}$, aromatic H$)$, and $8.15(2 \mathrm{H}, \mathrm{m}$, aromatic H$) ; \mathrm{m} / \mathrm{z}$ (E.I.) $281\left(M^{+}, 5 \%\right), 216\left(M^{+}-65,24\right)$, and $105\left(M^{+}-176\right.$, 100).
(1RS,2SR,6RS,7SR)-2-(4-Methoxybenzoyl)-4-methyl-3,5-di-oxo-4-azatricylo[5.2.1.0 ${ }^{2.6}$ ]dec-8-ene (20) ( $64 \%$ ), m.p. 151$152{ }^{\circ} \mathrm{C}$ (from ether-light petroleum) (Found: C, 69.55; H, 5.7; $\mathrm{N}, 4.3 . \mathrm{C}_{18} \mathrm{H}_{17} \mathrm{NO}_{4}$ requires $\mathrm{C}, 69.45 ; \mathrm{H}, 5.5 ; \mathrm{N}, 4.5 \%$ );
$v_{\text {max. }}\left(\mathrm{CH}_{2} \mathrm{Cl}_{2}\right) 1770$ and $1700(\mathrm{C}=\mathrm{O})$, and $1600 \mathrm{~cm}^{-1}$ (aromatic); $\delta_{\mathrm{H}}\left(\mathrm{CDCl}_{3}\right) 1.54$ and 1.70 (each $1 \mathrm{H}, \mathrm{d}, J 9.2 \mathrm{~Hz}$, $\mathrm{CH}_{2}$ ), 2.87 ( $3 \mathrm{H}, \mathrm{s}, \mathrm{NMe}$ ), 3.41 and 3.87 (each $1 \mathrm{H}, \mathrm{m}, 7-\mathrm{H}$ and 1H), $3.89(3 \mathrm{H}, \mathrm{s}, \mathrm{OMe}), 4.08(1 \mathrm{H}, \mathrm{d}, J 4.6 \mathrm{~Hz}, 6-\mathrm{H}), 6.28(2 \mathrm{H}, \mathrm{m}$, vinylic H), and 6.97 and 8.22 (each $2 \mathrm{H}, \mathrm{m}$, aromatic H ); $\mathrm{m} / \mathrm{z}$ (E.I.) $311\left(M^{+}, 5 \%\right)$ and $135\left(M^{+}-176,100\right)$.
(1RS,2SR,6RS,7SR)-2-(1-Oxopropyl)-4-methyl-3,5-dioxo-4azatricyclo[5.2.1.0 ${ }^{2,6}$ ]dec-8-ene (21) ( $30 \%$ ), a colourless oil (Found: $M^{+}, 233.1051 . \mathrm{C}_{13} \mathrm{H}_{15} \mathrm{NO}_{3}$ requires $M, 233.1052$ ); $\nu_{\text {max. }}\left(\mathrm{CH}_{2} \mathrm{Cl}_{2}\right) 1775$ and $1700 \mathrm{~cm}^{-1}(\mathrm{C}=\mathrm{O}) ; \delta_{\mathrm{H}}\left(\mathrm{CDCl}_{3}\right) 1.09(3$ $\left.\mathrm{H}, \mathrm{t}, J 7 \mathrm{~Hz}, \mathrm{CH}_{2} \mathrm{CH}_{3}\right), 1.58(1 \mathrm{H}$, br d, $J 9 \mathrm{~Hz}, 10-\mathrm{H}), 1.70(1 \mathrm{H}$, $\mathrm{dt}, J 9,1.5 \mathrm{~Hz}, 10-\mathrm{H}), 2.84(3 \mathrm{H}, \mathrm{s}, \mathrm{NMe}), 2.84$ and 2.92 (each 1 H , $\mathrm{dq}, J 19,7 \mathrm{~Hz}, \mathrm{CH}_{2} \mathrm{CH}_{3}$ ), 3.38 and 3.52 (each $1 \mathrm{H}, \mathrm{m}, 7-\mathrm{H}$ and 1H), $3.72(1 \mathrm{H}, \mathrm{d}, J 4.6 \mathrm{~Hz}, 6-\mathrm{H}$ ), and $6.19(2 \mathrm{H}, \mathrm{m}$, vinylic H$)$; $m / z$ (E.I.) $233\left(M^{+}, 14 \%\right), 177\left(M^{+}-56,33\right), 168\left(M^{+}-65,38\right)$, and $66\left(M^{+}-167,100\right)$.
(1RS,2SR,6RS,7SR)-4-Methyl-2-(2-methyl-1-oxopropyl)-3,5-dioxo-4-azatricyclo[5.2.1.0 ${ }^{2.6}$ ]dec-8-ene (22) (35\%), m.p. 63$64{ }^{\circ} \mathrm{C}$ (from ether-light petroleum) (Found: C, 67.9; H, 6.75; $\mathrm{N}, 5.7 . \mathrm{C}_{14} \mathrm{H}_{17} \mathrm{NO}_{3}$ requires $\mathrm{C}, 68.0 ; \mathrm{H}, 6.95 ; \mathrm{N}, 5.65 \%$ ); $v_{\text {max. }}\left(\mathrm{CH}_{2} \mathrm{Cl}_{2}\right) 3060$ (vinylic CH ), and 1775 and $1700 \mathrm{~cm}^{-1}$ $(\mathrm{C}=\mathrm{O}) ; \delta_{\mathrm{H}}\left(\mathrm{CDCl}_{3}\right) 1.08$ and 1.16 (each $3 \mathrm{H}, \mathrm{d}, J 6.7 \mathrm{~Hz}, \mathrm{CHMe}{ }_{2}$ ) $1.64(1 \mathrm{H}$, br d, $J 9 \mathrm{~Hz}, 10-\mathrm{H}), 1.71(1 \mathrm{H}, \mathrm{dt}, J 9,1.5 \mathrm{~Hz}, 10-\mathrm{H})$, $2.85(3 \mathrm{H}, \mathrm{s}, \mathrm{NMe}), 3.40(1 \mathrm{H}, \mathrm{m}, 7-\mathrm{H}), 3.49(1 \mathrm{H}, \mathrm{h}, J 6.7 \mathrm{~Hz}$, $\mathrm{CH} \mathrm{Me}_{2}$ ), $3.58(1 \mathrm{H}, \mathrm{m}, 1-\mathrm{H}), 3.71(1 \mathrm{H}, \mathrm{d}, J 4.6 \mathrm{~Hz}, 6-\mathrm{H})$, and 6.21 ( $2 \mathrm{H}, \mathrm{m}$, vinylic H); $m / z$ (E.I.) 247 ( $M^{+}, 11 \%$ ), 177 ( $M^{+}-$ $70,50), 66\left(M^{+}-181,84\right)$, and $43\left(M^{+}-244,100\right)$.
(1SR,2SR,5RS,6SR)-1-(4-Methylbenzoyl)-2,5,8-trimethyl-7,9-dioxo-8-azabicyclo[4.3.0]non-3-ene (23) (55\%), m.p. 110$112{ }^{\circ} \mathrm{C}$ (from ether-light petroleum) (Found: C, 73.3; H, 6.9; $\mathrm{N}, 4.65 . \mathrm{C}_{19} \mathrm{H}_{21} \mathrm{NO}_{3}$ requires $\mathrm{C}, 73.3 ; \mathrm{H}, 6.8 ; \mathrm{N}, 4.5 \%$ ); $v_{\text {max. }}\left(\mathrm{CH}_{2} \mathrm{Cl}_{2}\right) 3040($ aromatic CH$), 1775$ and $1705(\mathrm{C}=\mathrm{O})$, and $1610 \mathrm{~cm}^{-1}$ (aromatic ring); $\delta_{\mathrm{H}}\left(\mathrm{CDCl}_{3}\right) 1.29$ and 1.49 (each 3 H , d, $\left.J 7 \mathrm{~Hz}, 2 \times \mathrm{CHCH}_{3}\right), 2.39(3 \mathrm{H}, \mathrm{s}$, aromatic Me$), 2.55(1 \mathrm{H}$, $\mathrm{m}, 5-\mathrm{H}), 3.02(3 \mathrm{H}, \mathrm{s}, \mathrm{NMe}), 3.20(1 \mathrm{H}, \mathrm{d}, J 5.3 \mathrm{~Hz}, 6-\mathrm{H}), 3.32(1$ $\mathrm{H}, \mathrm{m}, 2-\mathrm{H}$ ), 5.68 and 5.73 (each $1 \mathrm{H}, \mathrm{dt}, J 9,3 \mathrm{~Hz}$, vinylic H), 7.22 ( $2 \mathrm{H}, \mathrm{m}$, aromatic H ), and $7.53(2 \mathrm{H}, \mathrm{m}$, aromatic H ); $m / z$ (E.I.) $311\left(M^{+}, 1 \%\right), 192\left(M^{+}-119,100\right), 119\left(M^{+}-192,100\right)$, and 91 ( $M^{+}-220,75$ ).
(1SR,2SR,5RS,6SR)-1-Benzoyl-2,5,8-trimethyl-7,9-dioxo-8-azabicyclo[4.3.0]non-3-ene (24) (71\%), m.p. 124- $127^{\circ} \mathrm{C}$ (from ether-light petroleum) (Found: $\mathrm{C}, 72.9 ; \mathrm{H}, 6.5 ; \mathrm{N}, 4.55$. $\mathrm{C}_{18} \mathrm{H}_{19} \mathrm{NO}_{3}$ requires $\mathrm{C}, 72.7 ; \mathrm{H}, 6.45 ; \mathrm{N}, 4.7 \%$ ); $v_{\text {max. }}\left(\mathrm{CH}_{2} \mathrm{Cl}_{2}\right)$ 1775 and $1705(\mathrm{C}=\mathrm{O})$, and $1600 \mathrm{~cm}^{-1}$ (aromatic ring); $\delta_{\mathrm{H}}\left(\mathrm{CDCl}_{3}\right) 1.3$ and $1.5\left(\right.$ each $\left.3 \mathrm{H}, \mathrm{d}, J 7.3 \mathrm{~Hz}, 2 \times \mathrm{CHCH}_{3}\right), 2.56$ $(1 \mathrm{H}, \mathrm{m}, 5-\mathrm{H}), 3.02(3 \mathrm{H}, \mathrm{s}, \mathrm{NMe}), 3.22(1 \mathrm{H}, \mathrm{d}, J 5 \mathrm{~Hz}, 6-\mathrm{H}), 3.22$ $(1 \mathrm{H}, \mathrm{m}, 2-\mathrm{H}), 5.68$ and 5.74 (each $1 \mathrm{H}, \mathrm{dt}, J 9,3 \mathrm{~Hz}$, vinylic H ), and $7.55\left(5 \mathrm{H}, \mathrm{m}\right.$, aromatic H); $m / z$ (C.I.) $298\left(M^{+}+1,81 \%\right.$ ), $192\left(M^{+}-105,100\right)$, and $105\left(M^{+}-192,47\right)$.
(1SR,2SR,5RS,6SR)-1-(4-Methoxybenzoyl)-2,5,8-trimethyl-7,9-dioxo-8-azabicyclo[4.3.0]non-3-ene (25) ( $52 \%$ ), m.p. 126$127^{\circ} \mathrm{C}$ (from ether); $v_{\text {max. }}\left(\mathrm{CH}_{2} \mathrm{Cl}_{2}\right) 1775$ and $1705(\mathrm{C}=\mathrm{O})$, and $1605 \mathrm{~cm}^{-1}$ (aromatic ring); $\delta_{\mathrm{H}}\left(\mathrm{CDCl}_{3}\right) 1.28(3 \mathrm{H}, \mathrm{d}, J 7 \mathrm{~Hz}$, $\left.\mathrm{CHCH}_{3}\right), 1.48\left(3 \mathrm{H}, \mathrm{d}, J 7.4 \mathrm{~Hz}, \mathrm{CHCH}_{3}\right), 2.53(1 \mathrm{H}, \mathrm{m}, 5-\mathrm{H})$, $3.02(3 \mathrm{H}, \mathrm{s}, \mathrm{NMe}), 3.20(1 \mathrm{H}, \mathrm{d}, J 5.2 \mathrm{~Hz}, 6-\mathrm{H}), 3.34(1 \mathrm{H}, \mathrm{m}, 2-$ H), $3.85(3 \mathrm{H}, \mathrm{s}, \mathrm{OMe}$ ), 5.67 and 5.72 (each $1 \mathrm{H}, \mathrm{dt}, J 9,3 \mathrm{~Hz}$, vinylic H), and 6.89 and 7.65 (each $2 \mathrm{H}, \mathrm{m}$, aromatic H); $m / z$ (E.I.) $327\left(M^{+}, 2 \%\right), 192\left(M^{+}-135,100\right)$, and $135\left(M^{+}-192\right.$, 98).
(1SR,2SR,5RS,6SR)-1-(1-Oxopropyl)-2,5,8-trimethyl-7,9-di-oxo-8-azabicyclo[4.3.0]non-3-ene (26) (12\%), m.p. $59-60^{\circ} \mathrm{C}$ (from ether-light petroleum) (Found: $M^{+}$, 249.1366. $\mathrm{C}_{14} \mathrm{H}_{19} \mathrm{NO}_{3}$ requires $M, 249.1364$ ); $v_{\text {max. }}\left(\mathrm{CH}_{2} \mathrm{Cl}_{2}\right) 1770$ and $1700 \mathrm{~cm}^{-1}(\mathrm{C}=\mathrm{O}) ; \delta_{\mathrm{H}}\left(\left[{ }^{2} \mathrm{H}_{6}\right]\right.$ benzene $) 0.88(3 \mathrm{H}, \mathrm{t}, J 7 \mathrm{~Hz}$, $\mathrm{CH}_{2} \mathrm{CH}_{3}$ ), 1.30 and 1.37 (each $3 \mathrm{H}, \mathrm{d}, J 7.3 \mathrm{~Hz}, 2 \times \mathrm{CHCH}_{3}$ ), $1.80\left(1 \mathrm{H}, \mathrm{dq}, J 19,7 \mathrm{~Hz}, \mathrm{HCHCH}_{3}\right), 2.13(1 \mathrm{H}, \mathrm{m}, 5-\mathrm{H}), 2.45(1$ $\mathrm{H}, \mathrm{dq} J 19,7 \mathrm{~Hz}, \mathrm{HCHCH}_{3}$ ), $2.53(1 \mathrm{H}, \mathrm{d}, J 5.7 \mathrm{~Hz}, 6-\mathrm{H}), 2.63$ ( 3 $\mathrm{H}, \mathrm{s}, \mathrm{NMe}), 2.92(1 \mathrm{H}, \mathrm{m}, 2-\mathrm{H})$, and 5.38 and 5.44 (each $1 \mathrm{H}, \mathrm{dt}, J$

9, 3 Hz , vinylic H); $m / z$ (E.I.) $249\left(M^{+}, 2 \%\right)$ and $192\left(M^{+}-57\right.$, 100).
(1SR,2SR,5RS,6SR)-1-(2-Methyl-1-oxopropyl)-2,5,8-tri-methyl-7,9-dioxo-8-azabicyclo[4.3.0]non-3-ene (27) ( $29 \%$ ), m.p. $73-76^{\circ} \mathrm{C}$ (from ether-light petroleum) (Found: C, $68.1 ; \mathrm{H}, 7.9$; $\mathrm{N}, 5.4 . \mathrm{C}_{15} \mathrm{H}_{21} \mathrm{NO}_{3}$ requires $\mathrm{C}, 68.4 ; \mathrm{H}, 8.05 ; \mathrm{N}, 5.25 \%$ ); $v_{\text {max. }}\left(\mathrm{CH}_{2} \mathrm{Cl}_{2}\right) 3050$ (vinylic CH ), and 1770 and $1700 \mathrm{~cm}^{-1}$ ( $\mathrm{C}=\mathrm{O}$ ); $\delta_{\mathrm{H}}$ ( $\left[{ }^{2} \mathrm{H}_{6}\right]$ benzene) $0.76\left(3 \mathrm{H}, \mathrm{d}, J 6.8 \mathrm{~Hz}, \mathrm{CHCH}_{3}\right), 1.04$ $\left(3 \mathrm{H}, \mathrm{d}, J 6.6 \mathrm{~Hz}, \mathrm{CHCH}_{3}\right), 1.31\left(3 \mathrm{H}, \mathrm{d}, J 7.2 \mathrm{~Hz}, \mathrm{CHCH}_{3}\right) 1.37$ ( $3 \mathrm{H}, \mathrm{d}, J 7.4 \mathrm{~Hz}, \mathrm{CHCH}_{3}$ ), $2.16(1 \mathrm{H}, \mathrm{m}, 5-\mathrm{H}), 2.62(3 \mathrm{H}, \mathrm{s}$, NMe), $2.72\left(1 \mathrm{H}, \mathrm{h}, J 6.7 \mathrm{~Hz}, \mathrm{C} H \mathrm{Me}_{2}\right), 2.75(1 \mathrm{H}, \mathrm{d}, J 5.7 \mathrm{~Hz}, 6-$ $\mathrm{H}), 2.94(1 \mathrm{H}, \mathrm{m}, 2-\mathrm{H})$, and 5.38 and 5.44 (each $1 \mathrm{H}, \mathrm{dt}, J 9,3 \mathrm{~Hz}$, vinylic H ); $m / z($ E.I. $) 264\left(M^{+}+1,1 \%\right)$, and $192\left(M^{+}-71\right.$, 100).

## Acknowledgements

We thank the S.E.R.C. for support (to M. L. D.), Mrs. E. McGuinness and Dr. A. Derome for the ${ }^{1} \mathrm{H}$ n.m.r. spectra, and Dr. R. T. Aplin for the mass spectra.

## References

1 H. K. Hall, Jr., P. Nogues, J. W. Rhoades, R. C. Sentman, and M. Detar, J. Org. Chem., 1982, 47, 1451.
2 S. J. Bailey, E. J. Thomas, S. M. Vather, and J. Wallis, J. Chem. Soc., Perkin Trans. 1, 1983, 851.
3 F. P. Hauck, Jr., and J. T. Fan, J. Org. Chem., 1969, 34, 1703.
4 H. J. Reich, J. M. Renga, and I. L. Reich, J. Am. Chem. Soc., 1975, 97, 5434.

4 L. M. Jackman and S. Sternhell, 'Applications of Nuclear Magnetic Resonance Spectroscopy in Organic Chemistry,' Pergamon, Oxford, 2nd edn., 1969, pp. 289 and 334.
6 A. P. Kozikowski, E. Hine, and J. P. Springer, J. Am. Chem. Soc., 1982, 104, 2059; M. Y. Kim, J. E. Starrett, Jr., and S. M. Weinreb, J. Org. Chem., 1981, 46, 5383.
7 D. Caine, C. R. Harrison, and D. G. Van Derveer, Tetrahedron Lett., 1983, 1353.
8 L. M. Rice, E. E. Reid, and C. H. Grogan, J. Org. Chem., 1954, 19, 884.
9 'Dictionary of Organic Compounds,' 5th edn., Chapman and Hall, New York, 1982, vol. 4, p. 4050.

